

Half Life

- time for 1/2 of the nuclei in a radioactive sample to decay

$$N = N_0 \left(\frac{1}{2}\right)^{t/T_{1/2}}$$

Ex.1 Radon 222 has a half life of 3.8 days. How long will it take to decay to 10%


Ex.2 ${}_{90}^{228}\text{Th}$ has a half life of 1.913 years as it alpha decays.

(a) Write the decay reaction.

(b) If you have a 2.5g sample, how long will it take for it to decay to only 1.0g?

Activity- the number of decays per second
- measured in Bequerel (Bq)

$$A = A_0 \left(\frac{1}{2} \right)^{t/T_{1/2}}$$

 http://www.colorado.edu/physics/2000/isotopes/radioactive_decay3.html

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Ex. 3 A 1000 Bq source with a half life of 15 hours is placed into a monitoring container. How long will it take the sample to decay to an activity of 125 Bq?

